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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
09/665,767	09/20/2000	James Claude Carnahan	RD-27,100	8695

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GENERAL ELECTRIC COMPANY  
GLOBAL RESEARCH CENTER  
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PO BOX 8, BLDG. K-1 ROSS  
NISKAYUNA, NY 12309

EXAMINER

CYGAN, MICHAEL T

ART UNIT

PAPER NUMBER

2856

DATE MAILED: 10/21/2002

Please find below and/or attached an Office communication concerning this application or proceeding.

<b>Office Action Summary</b>	Application No.	Applicant(s)
	09/665,767	CARNAHAN ET AL.
	Examiner Michael Cygan	Art Unit 2856

-- The MAILING DATE of this communication appears on the cover sheet with the correspondence address --

#### Period for Reply

A SHORTENED STATUTORY PERIOD FOR REPLY IS SET TO EXPIRE 3 MONTH(S) FROM THE MAILING DATE OF THIS COMMUNICATION.

- Extensions of time may be available under the provisions of 37 CFR 1.136(a). In no event, however, may a reply be timely filed after SIX (6) MONTHS from the mailing date of this communication.
- If the period for reply specified above is less than thirty (30) days, a reply within the statutory minimum of thirty (30) days will be considered timely.
- If NO period for reply is specified above, the maximum statutory period will apply and will expire SIX (6) MONTHS from the mailing date of this communication.
- Failure to reply within the set or extended period for reply will, by statute, cause the application to become ABANDONED (35 U.S.C. § 133).
- Any reply received by the Office later than three months after the mailing date of this communication, even if timely filed, may reduce any earned patent term adjustment. See 37 CFR 1.704(b).

#### Status

1) Responsive to communication(s) filed on 11 September 2002.

2a) This action is FINAL. 2b) This action is non-final.

3) Since this application is in condition for allowance except for formal matters, prosecution as to the merits is closed in accordance with the practice under *Ex parte Quayle*, 1935 C.D. 11, 453 O.G. 213.

#### Disposition of Claims

4) Claim(s) 1,4-29 and 31-36 is/are pending in the application.

4a) Of the above claim(s) \_\_\_\_\_ is/are withdrawn from consideration.

5) Claim(s) \_\_\_\_\_ is/are allowed.

6) Claim(s) 1,4-29 and 31-36 is/are rejected.

7) Claim(s) \_\_\_\_\_ is/are objected to.

8) Claim(s) \_\_\_\_\_ are subject to restriction and/or election requirement.

#### Application Papers

9) The specification is objected to by the Examiner.

10) The drawing(s) filed on \_\_\_\_\_ is/are: a) accepted or b) objected to by the Examiner.  
Applicant may not request that any objection to the drawing(s) be held in abeyance. See 37 CFR 1.85(a).

11) The proposed drawing correction filed on 08 February 2002 is: a) approved b) disapproved by the Examiner.  
If approved, corrected drawings are required in reply to this Office action.

12) The oath or declaration is objected to by the Examiner.

#### Priority under 35 U.S.C. §§ 119 and 120

13) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).

a) All b) Some \* c) None of:  
1. Certified copies of the priority documents have been received.  
2. Certified copies of the priority documents have been received in Application No. \_\_\_\_\_.  
3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).

\* See the attached detailed Office action for a list of the certified copies not received.

14) Acknowledgment is made of a claim for domestic priority under 35 U.S.C. § 119(e) (to a provisional application).

a) The translation of the foreign language provisional application has been received.

15) Acknowledgment is made of a claim for domestic priority under 35 U.S.C. §§ 120 and/or 121.

#### Attachment(s)

1) Notice of References Cited (PTO-892)

2) Notice of Draftsperson's Patent Drawing Review (PTO-948)

3) Information Disclosure Statement(s) (PTO-1449) Paper No(s) \_\_\_\_\_

4) Interview Summary (PTO-413) Paper No(s) \_\_\_\_\_.  
5) Notice of Informal Patent Application (PTO-152)  
6) Other: \_\_\_\_\_

## DETAILED ACTION

### ***Continued Examination Under 37 CFR 1.114***

1. A request for continued examination under 37 CFR 1.114, including the fee set forth in 37 CFR 1.17(e), was filed in this application after final rejection. Since this application is eligible for continued examination under 37 CFR 1.114, and the fee set forth in 37 CFR 1.17(e) has been timely paid, the finality of the previous Office action has been withdrawn pursuant to 37 CFR 1.114. Applicant's submission filed on 22 July 2002 has been entered.

### ***Claim Objections***

2. Claims 9 and 10 are objected to because claim 9 does not further limit parent claim 1 since the added limitation is inherent to claim 1 (i.e., the sample comprising a polymerization reaction mixture). Appropriate correction is required.

### ***Claim Rejections - 35 USC § 103***

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

3. Claims 1, 4-11, 18, 19, 21-24, 26-29, and 31-36 are rejected under 35 U.S.C. 103(a) as being unpatentable over Miroslav (US 6,296,771 B1) in view of Allcock ("Contemporary Polymer Chemistry", 1990). Miroslav discloses an analysis system and method for polymer weight determination which comprises injecting a known amount of sample into an analysis system containing a GPC (size-exclusive; see column 1, lines 48-49; column 14, lines 52-64; column 18, lines 47-48) column, an in-line concentration detector and a molar mass detector (such as a differential refractive index detector and a light scattering detector, see column 20, lines 26-39); wherein a high molecular weight fraction is separated with minimal dispersion, analyzed to determine concentration and molar mass, and an average molecular weight derived therefrom (column 21, lines 1-22; average molecular weights are inherently calculated from concentration and molecular mass). The average molecular weight may be number-averaged or weight averaged; see column 21, lines 4-16. The total analysis time may be 60 seconds (column 12, lines 33-36). A plurality of samples are provided from a sample preparation array (Figure 5) and analysis is conducted automatically with an autoinjector (column 12, lines 1-25), a solvent preparation and delivery system (Figure 3 and description at columns 7-11, particularly column 7, lines 47-49), a chromatographic column [102], detectors [103] (such as a differential

refractive index detector and a light scattering detector, see column 20, lines 26-39), and a computer [222]. See entire document.

Miroslav teaches the claimed invention except for the sample being the product of a polymer reaction between a diphenyl carbonate and a dihydric phenol. With respect to the sample being the product of a polymer reaction between a diphenyl carbonate and a dihydric phenol, Miroslav teaches only that the disclosed invention is "for characterizing combinatorial libraries of material samples such as polymer samples, and particularly, libraries of or derived from reaction mixtures such as polymerization product mixtures, to facilitate the discovery of commercially important materials". Allcock teaches that polycarbonates "of particular importance" are formed by reaction of 2,2-bis(4-hydroxyphenyl)propane (a dihydric phenol) and diphenyl carbonate, (and inherently, an appropriate solvent) see page 29. It would have been obvious to one having ordinary skill in the art at the time the invention was made to use a product of a polymer reaction between a diphenyl carbonate and a dihydric phenol as taught by Allcock in the invention of Miroslav as a sample for analysis in order to provide advantageous use of the invention of Miroslav, since Allcock teaches that such a product is "of particular importance", and Miroslav states that his invention is to be used with polymerization reaction product mixtures having commercial importance.

With respect to claims 6-8, Miroslav discloses the sample containing a solvent chosen from a group comprising “typical solvents” such as tetrahydrofuran or toluene.

With respect to claim 21, Miroslav discloses serial (sequential) detection at column 20, lines 23-39.

With respect to claims 22 and 33, while Miroslav discloses online techniques, the examiner takes Official Notice of the equivalence of offline and online techniques in the analysis art, and the use of either of these techniques would have been obvious to one having ordinary skill in the art at the time the invention was made.

4. Claims 12-17, 20, and 25 are rejected under 35 U.S.C. 103(a) as being unpatentable over Miroslav (US 6,296,771 B1) in view of Allcock (“Contemporary Polymer Chemistry”, 1990), further in view of Nielsen (US 6,175,409 B1). With respect to claims 12-17 and 25, the claims are considered to be met by Miroslav in view of Allcock except for an analysis time less than 40, 30, 20, 10, 5, or 3 seconds. With respect to the analysis time, Nielsen discloses the total analysis time for the above-disclosed system (showing minimal dispersion; see Figure 8) and method to be “not more than 1 second per sample” for determining average molecular weight if desired; see column 23 through column 24; see also columns 40-41. Since Miroslav is directed toward the same system and

method of Nielsen, it would have been obvious to one having ordinary skill in the art at the time the invention was made to use the time structure of Nielsen in the invention of Miroslav to achieve the speed of analysis as disclosed by Nielsen since this would result in a rapid analysis capable of a high sample throughput. Note that the inventions of Miroslav and Nielsen both derive priority from the same invention disclosed in a provisional application (60/157,338).

With respect to claim 20, Miroslav discloses a dynamic (i.e., operating at multiple angles) light scattering detector (column 20, lines 26-32), but does not disclose a step of utilizing multiple angles for detection. Nielsen discloses that multiple angles can be used for light scattering measurements, see Figure 8. It would have been obvious to one having ordinary skill in the art at the time the invention was made to use multiple angles for detection as taught by Nielsen in the invention of Miroslav, since detection at multiple angles with a dynamic light scattering detector provides more information as to the nature of the sample resulting in a more accurate analysis. Note that the inventions of Miroslav and Nielsen both derive priority from the same invention disclosed in a provisional application (60/157,338).

### ***Response to Arguments***

5. Applicant's arguments with respect to the claims have been considered but are moot in view of the new ground(s) of rejection.

6. The statement regarding online/offline techniques as common knowledge is taken to be admitted prior art because applicant failed to traverse the examiner's assertion of Official Notice in any prior response.

### ***Conclusion***

7. The prior art made of record and not relied upon is considered pertinent to applicant's disclosure. Ogoe (US 6,329,450 B1) discloses determination of the weight average molecular weight of "useful" polycarbonates derived from reactions of diphenyl carbonate and 2,2-bis(4-hydroxyphenyl)propane (a dihydric phenol which is also known as bisphenol) through light scattering or gel permeation chromatography. Watanabe (US 5,414,144) discloses determination of properties of polycarbonates derived from reactions of diphenyl carbonate and 2,2-bis(4-hydroxyphenyl)propane (a dihydric phenol which is also known as bisphenol) through infrared adsorption or chromatography. Fukawa (US 5,214,073) discloses useful applications of polycarbonates derived from reactions of diphenyl carbonate and 2,2-bis(4-hydroxyphenyl)propane (a dihydric phenol which is also known as bisphenol).

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Michael Cygan whose telephone number is 703-305-0846. The examiner can normally be reached on 8:30-6 M-Th, alternate Fridays.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Hezron Williams can be reached on 703-305-4705. The fax phone numbers for the organization where this application or proceeding is assigned are 703-308-7722 for regular communications and 703-308-7722 for After Final communications.

Any inquiry of a general nature or relating to the status of this application or proceeding should be directed to the receptionist whose telephone number is 703-306-3431.



Michael Cygan, Ph.D.  
October 17, 2002